

CLASS XII

DATE : 27-06-2010

PAPER - II

ANSWER KEY WITH SOLUTION

MATHEMATICS

SECTION - A

1. B 2. B 3. C 4. A 5. C 6. C 7. A
8. C

SECTION - B

1. A-R ; B-R ; C-R 2. A-P, Q, R ; B-S, T ; C-T ; D-T

SECTION - C

- 1.0010 2.0006 3.0180 4.0001 5.0005 6.0003

PHYSICS

SECTION - A

1. A 2. D 3. C 4. A 5. A 6. A 7. A 8. B

SECTION - B

1. (A) → P ; (B) → R ; (C) → S 2. (A) → P, Q, S ; (B) → Q, R ; (C) → P, Q, S ; (D) → P, Q, S, T

SECTION - C

1. 0002 2. 0004 3. 0002 4. 0006 5. 0080 6. 0001

CHEMISTRY

SECTION - A

1. B 2. A 3. A 4. C 5. B 6. C 7. A
8. A

SECTION B

1. A-P, B-R, C-Q 2. A-P B-Q, R C-P, D-Q, R

SECTION C

1. 0035 2. 0153 3. 0001 4. 0006 5. 0005 6. 0010

SOLUTIONS

MATHEMATICS

SECTION – A

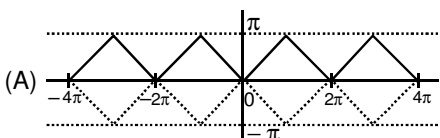
1. **B**
 $y = f(x) \Rightarrow f(1) = 5 \Rightarrow y = 5$ when $x = 1$
 $x = g(y)$ [g(y) \rightarrow inverse of f(x)]
 $1 = g'(y) \cdot f'(x) \Rightarrow g''(y) = -\frac{f''(x)}{[f'(x)]^3}$
 $\Rightarrow g''(5) = \frac{-4}{(2)^3} = \frac{-1}{2}$ (put $x = 1, y = 5$)
2. **B**
 Let three terms of G.P. be $\frac{a}{r}, a, ar$
 $a^3 = -8 \Rightarrow a = -2$
 also $-\frac{2}{r} - 2 - 2r = \frac{21}{10} \Rightarrow r = -\frac{4}{5}$ or $-\frac{5}{4}$ (reject)

Now $S_\infty = \frac{a/r}{1-r} = \frac{5/2}{1+4/5} = \frac{25}{18} = \frac{p}{q}$

$\therefore p - q = 7$

3. **C**
 Only possibility is
 $\sin^{-1} x_1 = \sin^{-1} x_2 = \dots = \sin^{-1} x_n = -\pi/2$
 $\Rightarrow x_1 = x_2 = \dots = x_n = -1$
4. **A**
5. **C**
 $f'(x) \cdot g(x) + f(x) \cdot g'(x) = 0 \Rightarrow \frac{f'(x)}{f(x)} = -\frac{g'(x)}{g(x)}$ (1)
 $f''(x) \cdot g(x) + 2f'(x) \cdot g'(x) + f(x) \cdot g''(x) = 0$ (2)
 $f'''(x) \cdot g(x) + 3f''(x) \cdot g'(x) + 3f'(x) \cdot g''(x) + f(x) \cdot g'''(x) = 0$ (3)
 Now $\frac{g(x)}{g'(x)} \left[\frac{f'''(x)}{f'(x)} - \frac{g'''(x)}{g'(x)} \right]$
 $= -3 \left[\frac{f''(x)}{f(x)} - \frac{g''(x)}{g(x)} \right] \frac{f(x)}{f'(x)} = 3 \left[\frac{f''(x)}{f(x)} - \frac{g''(x)}{g(x)} \right]$

SECTION – B

1. (A)–R ; (B)–R ; (C)–R
- 
- (A)
- (B) $g'(y) = \frac{1}{f'(x)}$
 $(x = g(y) \Rightarrow 2 = g(a)$ when $y = a, x = 2)$
 $g'(a) = \frac{1}{f'(2)} = 4$

(C) $L = \lim_{x \rightarrow 0} \frac{(e^{2x} + 1)(e^x - x - 1)}{e^x (e^x - 1) \cdot x}$
 $= \lim_{x \rightarrow 0} \frac{(e^{2x} + 1)(e^x - x - 1)}{e^x \left(\frac{e^x - 1}{x} \right) x^2} = 1$

2. (A)–P, Q, R ; (B)–P, Q, R, S, T ; (C)–T ; (D)–T
- (B) $g(x) = x(x - 1)(x + 1)$
 $h(x) = \text{gof}(x) = \text{sgn } x(\text{sgn } x + 1)(\text{sgn } x - 1)$
- $$h(x) = \begin{cases} 0 & , x > 0 \\ 0 & , x = 0 \\ 0 & , x < 0 \end{cases}$$
- $h(x) = 0$ for all $x \in R$

(C) $L = \lim_{n \rightarrow \infty} \frac{\left\{ (n+1) \left(n + \frac{1}{2} \right) \dots \left(n + \frac{1}{2^{n-1}} \right) \right\}^n}{n^{n^2}}$

$L = \lim_{n \rightarrow \infty} \left\{ \frac{(n+1) \left(n + \frac{1}{2} \right) \dots \left(n + \frac{1}{2^{n-1}} \right)}{n^n} \right\}^n$

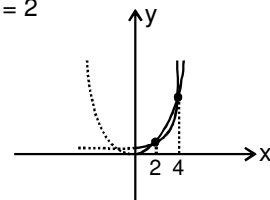
$L = \left\{ \left(1 + \frac{1}{n} \right) \left(1 + \frac{1}{2n} \right) \dots \left(1 + \frac{1}{2^{n-1}n} \right) \right\}^n$

$\ln(L) = \lim_{n \rightarrow \infty} n \left\{ \ln \left(1 + \frac{1}{n} \right) + \ln \left(1 + \frac{1}{2n} \right) + \dots + \ln \left(1 + \frac{1}{2^{n-1}n} \right) \right\}$
 $= \lim_{n \rightarrow \infty} \left\{ \frac{\ln \left(1 + \frac{1}{n} \right)}{1/n} + \frac{\ln \left(1 + \frac{1}{2n} \right)}{2 \cdot (1/2n)} + \dots + \frac{\ln \left(1 + \frac{1}{2^{n-1}n} \right)}{2^{n-1} \cdot (1/n2^{n-1})} \right\}$

$= 1 + \frac{1}{2} + \frac{1}{2^2} + \dots + \frac{1}{2^{n-1}}$

$= \left(\frac{1 - \frac{1}{2^n}}{1 - \frac{1}{2}} \right) = 2 \left(1 - \frac{1}{2^n} \right) = 2$

- (D) $2 \ln x = x \ln 2$
 $\Rightarrow x^2 = 2^x$
 $\Rightarrow x = 2, 4$
 \therefore two solution



SECTION – C

1. **0010**

$$x \in [-1, 0]$$

$$x + \frac{1+x^2}{2} = -2x \Rightarrow x^2 + 6x + 1 = 0$$

$$\Rightarrow x = \frac{-6 \pm \sqrt{36-4}}{2} = -3 \pm 2\sqrt{2}$$

$$\therefore x = 2\sqrt{2} - 3 \Rightarrow |10a| = |20\sqrt{2} - 30| \text{ \& } x \in [0, 1]$$

$$x + \frac{1+x^2}{2} = 2x \Rightarrow x^2 - 2x + 1 = 0 \Rightarrow (x-1)^2 = 0$$

$$\therefore x = 1 \Rightarrow |10a| = 10$$

$$|10a| = 10, (20\sqrt{2} - 30) \Rightarrow [|10a|] = 1, 10$$

2. **0007**

$$f(x) = \frac{[x] \cdot \sin \frac{x}{[x+1]}}{1+[x]} \quad (\because \sin \pi[x+1] = 0)$$

put $x = I + h$ where $I \rightarrow$ integers

$$\lim_{h \rightarrow 0} f(I+h) = \lim_{h \rightarrow 0} \frac{I \cdot \sin \frac{I+h}{I+1}}{I+1} = \frac{I \cdot \sin \frac{I}{I+1}}{I+1}$$

$$\lim_{h \rightarrow 0} f(I-h) = \frac{(I-1) \sin \frac{I-h}{I}}{I} = \frac{(I-1) \sin 1}{I}$$

\therefore Discontinuous function at all integers.

3. **0180**

$$f(x) = \begin{cases} ax^3 + b & , 0 \leq x \leq 1 \\ 2\cos \pi x + \tan^{-1} x & , 1 < x \leq 2 \end{cases}$$

$$f'(x) = \begin{cases} 3ax^2 & , 0 < x < 1 \\ -2\pi \sin \pi x + \frac{1}{1+x^2} & , 1 < x < 2 \end{cases}$$

As the function is differentiable in $[0, 2]$

\Rightarrow function is differentiable at $x = 1$

$$\therefore f'(1^-) = f'(1^+)$$

$$\Rightarrow 3a = \frac{1}{2} \Rightarrow a = \frac{1}{6}$$

Function will also be continuous at $x = 1$

$$\therefore \lim_{x \rightarrow 1^-} f(x) = \lim_{x \rightarrow 1^+} f(x)$$

$$\Rightarrow a + b = -2 + \frac{\pi}{4}$$

$$\therefore b = -2 - \frac{1}{6} + \frac{\pi}{4} = \frac{\pi}{4} - \frac{13}{6}$$

$$\Rightarrow k_1 = 6 \text{ and } k_2 = 12 \Rightarrow k_1^2 + k_2^2 = 180$$

4. **0001**

$$L_1 = \lim_{x \rightarrow 0^-} (\cos x - \sin x)^{\operatorname{cosec} x} = e^\ell$$

$$\ell = \lim_{x \rightarrow 0^-} \left(\frac{\cos x - \sin x - 1}{\sin x} \right)$$

$$= \lim_{x \rightarrow 0^-} \left(\frac{1 - 2\sin^2 \frac{x}{2} - 2\sin \frac{x}{2} \cdot \cos \frac{x}{2} - 1}{2\sin \frac{x}{2} \cdot \cos \frac{x}{2}} \right)$$

$$= - \lim_{x \rightarrow 0^-} \left(\frac{\sin \frac{x}{2} + \cos \frac{x}{2}}{\cos \frac{x}{2}} \right) = -1$$

$\Rightarrow L_1 = e^{-1} \therefore a = 1/e$ (as function is continuous)

$$L_2 = \lim_{x \rightarrow 0^+} \frac{e^{1/x} + e^{2/x} + e^{3/x}}{a \cdot e^{2/x} + b \cdot e^{3/x}}$$

Divided N^r & D^r by $e^{3/x}$

$$= \lim_{x \rightarrow 0^+} \frac{e^{-2/x} + e^{-1/x} + 1}{a \cdot e^{-1/x} + b} = \frac{1}{b}$$

$$\Rightarrow L_2 = \frac{1}{b} \text{ by definition } \frac{1}{b} = \frac{1}{e} \Rightarrow b = e$$

$$\therefore a \cdot b = 1$$

5. **0005**

$$y = e^{(\sin^2 x + \sin^4 x + \dots \infty) \log_e 2} = e^{\frac{\sin^2 x \cdot \log_e 2}{1 - \sin^2 x}}$$

$$= e^{(\tan^2 x) \cdot \log_e 2} = 2^{\tan^2 x}$$

roots of $x^2 - 9x + 8 = 0$ are 8 and 1

$$\Rightarrow 2^{\tan^2 x} = 2^3 \Rightarrow \tan x = \sqrt{3}$$

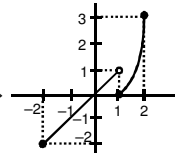
$$\text{and } 2^{\tan^2 x} = 2^0 \Rightarrow \tan x = 0 \text{ (reject)}$$

$$\therefore \frac{\sin x + \cos x}{\sin x - \cos x} = 2 + \sqrt{3} \Rightarrow a + b = 5$$

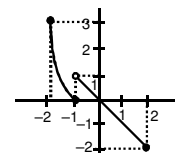
6. **0003**

$$g(x) = |f(x)| - f(-x)$$

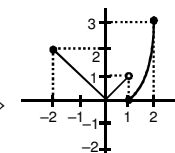
$$f(x) = \begin{cases} x & ; -2 \leq x < 1 \\ x^2 - 1 & ; 1 \leq x \leq 2 \end{cases}$$



$$f(-x) = \begin{cases} -x & ; -2 \leq -x < 1 \\ x^2 - 1 & ; 1 \leq -x \leq 2 \end{cases}$$



$$= \begin{cases} -x & ; -1 < x \leq 2 \\ x^2 - 1 & ; -2 \leq x \leq -1 \end{cases}$$



$$|f(x)| = \begin{cases} -x & ; -2 \leq x < 0 \\ x & ; 0 \leq x < 1 \\ x^2 - 1 & ; 1 \leq x \leq 2 \end{cases}$$

$$g(x) = \begin{cases} -x - x^2 + 1 & ; -2 \leq x \leq -1 \\ 0 & ; -1 \leq x < 0 \\ 2x & ; 0 \leq x < 1 \\ x^2 + x - 1 & ; 1 \leq x < 2 \end{cases}$$

Not differentiable at $-1, 0, 1$.

PHYSICS

SECTION - (A)

1. **A**
 $A < C \Rightarrow \sin 60^\circ < \sin C$
 $\frac{\sqrt{3}}{2} < \frac{1}{\mu} \Rightarrow \mu < \frac{2}{\sqrt{3}}$
2. **C**
 $(\Delta x)_p = 2y = \frac{\lambda}{2} \Rightarrow y = \frac{\lambda}{4}$
 $(\Delta x)_Q = 2(y - d) = \frac{3\lambda}{2}$
 $(\Delta x)_p = 2(y - 2d) = \frac{5\lambda}{2}$
 $\left| 2\left(\frac{\lambda}{4} - d\right) \right| = \frac{3\lambda}{2}$
 $2d = \lambda$
 $v = f\lambda \Rightarrow v = 2df$
3. **C**
4. **A**
5. **A**
6. **A**
 E will be zero at distance a, only when $Q_1 > Q_2$
7. **A**
 $\frac{kQ_1}{(a+l)^2} = \frac{kQ_2}{a^2} \Rightarrow \left| \frac{Q_1}{Q_2} \right| = \left(\frac{a+l}{a} \right)^2$
8. **B**
 E_{net} at b distance
 $E_b = \frac{kQ_1}{(l+b)^2} - \frac{kQ_2}{b^2}$
 $\frac{dE_b}{db} = 0 \Rightarrow b = \frac{l}{\left(\frac{Q_1}{Q_2}\right)^{1/3} - 1}$

SECTION - (B)

1. **(A) → P ; (B) → R ; (C) → S**
Case - I : when $mg \sin \theta < \mu mg \cos \theta$
 $f = mg \sin \theta$ As $\theta \uparrow f \uparrow$
 Static friction is working
Case - II When $mg \sin \theta > \mu mg \cos \theta$
 kinetic friction will act
 and $\theta \uparrow f \uparrow$

- (B)** $N = mg \cos \theta$
 $\theta = 0, N = mg$
 $\theta \uparrow, N \downarrow$
- (C) Case - I**
 $f_c = \sqrt{N^2 + (f_r)^2} = \sqrt{(mg \cos \theta)^2 + (mg \sin \theta)^2} = mg$
- Case - II**
 $f = \sqrt{(mg \cos \theta)^2 + \mu(mg \cos \theta)^2} = mg \cos \theta \sqrt{1 + \mu^2}$

2. **(A) → P, Q, S ; (B) → R ; (C) → P, Q, S ; (D) → P, Q, S, T**

SECTION - (C)

1. **2**
 $y = [3 \sin(kx + \omega t) + 3 \sin(kx - \omega t)] + 2 \sin(kx - \omega t)$
 $\downarrow \qquad \qquad \qquad \downarrow$
 Standing wave Traveling wave
2. **4**
 $n \cdot \lambda = 2\mu_2 t \cos r + \frac{\lambda}{2}$
3. **2**
 Since $T_0 = 2\pi \sqrt{\frac{l_0}{C}}$
 and $T = 2\pi \sqrt{\frac{l}{C}}$, where $l = l_0 + l_m$
 We have $l_m = \frac{C}{4\pi^2} (T^2 - T_0^2)$
 $= \frac{12.5}{4 \times 9.8} [(3.2)^2 - 2^2] = 2 \text{ kg m}^2$

4. **6**
-
- Power incident on slit 1
 $P_{s1} = I_0 \cdot 2A$
 Power incident on slit 2
 $P_{s2} = I_0 \cdot A$
 Total incident power = $P_1 = I_0 \cdot 2A + I_0 \cdot A = 3 I_0 A$
 Path difference = $(\mu - 1) t = \frac{\lambda}{8}$

Power goes in original direction

$$P_{s_1}' = 0.2 I_0 A$$

$$P_{s_2}' = 0.1 I_0 A$$

$$A_{\text{net}}^2 = A_1^2 + A_2^2 + 2A_1 A_2 \cos \phi$$

$$P = kA^2$$

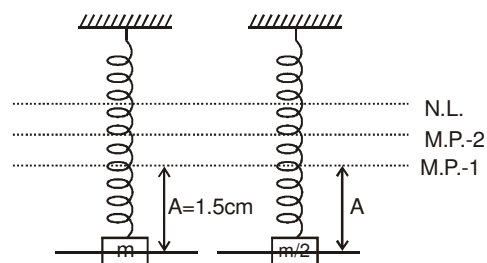
$$\Rightarrow P_{\text{net}} = P_{s_1}' + P_{s_2}' + 2\sqrt{P_{s_1}' P_{s_2}'} \cos\left(\frac{\lambda}{8} \times \frac{2\pi}{\lambda}\right)$$

$$P_2 = P_{\text{net}} = 0.2 I_0 A + 0.1 I_0 A + \sqrt{2} + \sqrt{0.2} I_0 A$$

$$P_2 = 0.5 I_0 A$$

$$\Rightarrow \frac{P_1}{P_2} = \frac{3I_0 A}{0.5I_0 A} = 6$$

5. 80



$$A_f = A + \frac{mg}{2k} = \left[1.5 \times 10^{-2} + \frac{5 \times 10}{2 \times 1000} \right] = 0.04 \text{ m}$$

\Rightarrow maximum speed = $A_f \omega$

$$= \left[\left(\sqrt{\frac{1000 \times 2}{5}} \right) \times 0.04 \right] = 80 \text{ cm/sec}$$

6. 1

At maximum stretched condition

$$(QE) x = \frac{1}{2} kx^2$$

$$x = \frac{2QE}{K} = \frac{2 \times 50 \times 10^{-6} \times 10^6}{100} = 1 \text{ m}$$

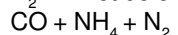
CHEMISTRY

SECTION - A

1.

B

N_2 will not be oxidised

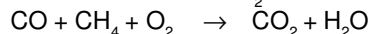


$$x \quad y \quad 20 - x - y$$

contraction = $x + y + \text{volume of } O_2 - (\text{vol. of } CO_2 + \text{vol. of unreacted } O_2)$

$$= x + y + \text{vol. of reacted } O_2 - 14$$

$$\Rightarrow \text{volume if reacted } O_2 = 27 - x - y$$



$$x \quad y \quad 27 - x - y \quad 14$$

P_{OAC} for C

$$x + y = 14 \quad \dots\dots\dots(1)$$

for H

$$4y = 2x \text{ moles of } H_2O$$

$$\Rightarrow \text{moles of } H_2O = 2y$$

P_{OAC} for O

$$x + 2(27 - x - y) = 2 \times 14 + \text{moles of } H_2O$$

$$x + 54 - 2x - 2y = 28 + 2y$$

$$x + 4y = 26 \quad \dots\dots\dots(ii)$$

from (i) & (ii)

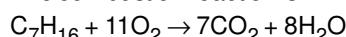
$$y = 4, x = 10$$

$$\text{vol. of } N_2 = 20 - x - y = 6$$

2.

A

The combustion reaction is



$$\Rightarrow -4853.40 = -7 \times 393.5 - 8 \times 285.9 - \Delta H_f^\circ$$

$$\Delta H_f^\circ = -188.3 \text{ kJ}$$

3.

A

$$\text{Given, } \frac{p}{V} = K \text{ (constant) Also, } \frac{p_1 V_1}{T_1} = \frac{p_2 V_2}{T_2}$$

$$\Rightarrow T_2 - T_1 \left(\frac{p_2 V_2}{p_1 V_1} \right) = T_1 \left(\frac{p_2}{p_1} \right)^2 = 4T_1$$

$$\Rightarrow \Delta U = C_v \Delta t = 3C_v T_1$$

$$\Rightarrow -dw = pdV = kVdV$$

$$\Rightarrow -w = \frac{k}{2} (V_2^2 - V_1^2): \text{ Also } \frac{p}{V} = k = \frac{RT}{V^2}$$

$$V^2 = \frac{RT}{K}$$

$$\Rightarrow V_2^2 - V_1^2 = \frac{R}{K} (T_2 - T_1) = \frac{3T_1 R}{K}$$

$$\Rightarrow w = -\frac{K}{2} \times \frac{3T_1 R}{K} = -\frac{3RT_1}{2}$$

$$\Rightarrow \frac{\Delta U}{w} = -\frac{2C_v}{R} = -3$$

4. **C**
 The change occur in following sequence:
 $\text{Ice}(-20^\circ\text{C}) \xrightarrow{q_1} \text{Ice}(0^\circ\text{C}) \xrightarrow{q_2} \text{H}_2\text{O}(l)(0^\circ\text{C})$
 $\xrightarrow{q_3} \text{H}_2\text{O}(l)(0^\circ\text{C})$
 $q_1 = 25 \times 2.03 \times 20 = 1015 \text{ J}$
 $q_2 = 333 \times 25 = 8325 \text{ J}$
 $q_3 = 15000 - 1015 - 8325 = 5660$
 $\Rightarrow 5660 = 25 \times 4.184 \times \Delta T$
 $\Rightarrow \Delta T = 54 \Rightarrow \text{Final temperature} = 54^\circ\text{C}$

5. **B**
 Number of moles of $\text{O}_2 = \frac{3}{4}$
 No of moles of $\text{H}_2\text{O}_2 = 1.5$
 $\% \text{ w/v} = \frac{1.5 \times 34}{500} \times 100 = 10.2 \%$

6. **C**
 For AB
 $\frac{P_0}{T_0} = \frac{2P_0}{T_B} \Rightarrow T_B = 2T_0$
 Also, $\frac{V_0}{2T_0} = \frac{2V_0}{T_C} \Rightarrow T_C = 4T_0$
 For AC (Isothermal Process)
 $\frac{P_0}{V_0} = \frac{2P_0}{V_C}$
 $V_C = 2V_0$

7. **A**
 Total work done = Area under curve
 $= -\frac{1}{2} \times P_0 \times V_0 = -\frac{1}{2} P_0 V_0$ (clockwise)
 Now $(\Delta U)_{\text{cycle}} = nC_V \Delta T = 0 = w + q_T$
 $q_T = \frac{1}{2} P_0 V_0$
 $q_{AB} = \Delta U_{AB} = n C_V \Delta T = \frac{P_0 V_0}{RT_0} \times \frac{5}{2} R (2T_0 - T_0)$
 $= \frac{5}{2} P_0 V_0$
 $q_{BC} = \Delta U_{BC} - w_{BC} = n C_V \Delta T + 2P_0 V_0$
 $= \frac{P_0 V_0}{RT_0} \times \frac{5}{2} R (4T_0 - 2T_0) + 2P_0 V_0$
 $= 5P_0 V_0 + 2P_0 V_0 = 7P_0 V_0$
 $q_{CA} = n C_V \Delta T - w_{CA} = 0 - \frac{3}{2} P_0 V_0 = -\frac{3}{2} P_0 V_0$
 $q_{AB} + q_{BC} = \left(\frac{5}{2} + 7\right) P_0 V_0 = \frac{19}{2} P_0 V_0$

8. **A**
 $\eta = \frac{|\text{work}|}{\text{Heat added}} = \frac{\frac{1}{2} P_0 V_0}{\frac{19 P_0 V_0}{2}} = \frac{100}{19} \%$

SECTION B
 1. **A-P, B-R, C-S**
 2. **A-Q, T B-Q, R C-P, D-Q, T**

SECTION C
 1. **35**
 2. **153**
 3. **1M**

No. of moles of $\text{AgNO}_3 = \frac{100 \times 8 \times 17}{100 \times 170} = 0.8 \text{ mole}$
 $\text{NaCl} \rightarrow \text{Na}^+ + \text{Cl}^-$
 $\quad \quad \quad 0.6 \quad 0.6$
 $\text{AgNO}_3 \rightarrow \text{Ag}^+ + \text{NO}_3^-$
 $\quad \quad \quad 0.8 \quad 0.8$
 $\quad \quad \quad \text{Cl}^- + \text{Ag}^+ \rightarrow \text{AgCl} \downarrow$
 t = 0 $\quad 0.6 \quad 0.8$
 After reaction $\quad 0 \quad 0.2 \quad 0.6$
 $[\text{M}]_{\text{Ag}^+} = \frac{0.2}{200 \times 10^{-3}} = 1\text{M}$

4. **6 litre**
 22.4 l of C_4H_{10} produces to 2800 kJ
 1.5 $\frac{2800}{22.4} \times 1.5$
 22.4 l of CH_4 produces to 700 kJ
 x $\frac{700}{22.4} \times x$
 $\Rightarrow \frac{2800}{22.4} \times 1.5 = \frac{700}{22.4} \times x$
 $\Rightarrow x = 6 \text{ l}$

5. **5**
 $\Delta U = \int n C_V dt = \int_{300}^{400} 2.5 \times (16.5 + 10^{-2} T) dT$
 $= 5000 \text{ J} = 5 \text{ kJ}$

6. **11**