

**CLASS XI**

**DATE : 13-06-2010**

**ANSWER KEY WITH SOLUTION**

**MATHEMATICS**

*SECTION - A*

1. B    2. A    3. B    4. C    5. B    6. B    7. B  
8. B    9. B,C    10. A,C    11. B,C    12. B    13. C    14. A  
15. C

*SECTION - C*

1. 0002    2. 0006    3. 0004    4. 0003    5. 0001

**PHYSICS**

*SECTION - A*

1. A    2. D    3. B    4. C    5. C    6. A    7. A  
8. B    9. A,B,C,D    10. A,B,D    11. B,D    12. A    13. C    14. B  
15. D

*SECTION - C*

1. 0150    2. 0010    3. 0006    4. 0006    5. 0007

**CHEMISTRY**

*SECTION A*

1. C    2. A    3. A    4. B    5. C    6. B    7. C  
8. C    9. ABCD    10. ABC    11. ABC    12. B    13. B    14. B  
15. A

*SECTION C*

1. 0050    2. 0016    3. 2200    4. 0016    5. 4000

# PHYSICS

## SECTION - A

1. **A**  
By sine Rule :

$$\frac{\sin 53^\circ}{2x^2 - 1} = \frac{\sin 90^\circ}{\frac{5}{4}x} = \frac{\sin 37^\circ}{\frac{3}{4}y}$$

$$\Rightarrow x = y$$

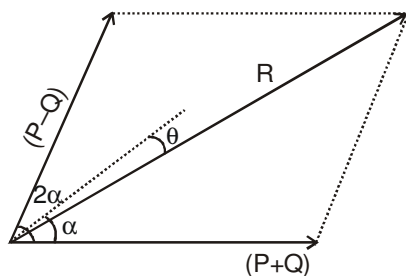
2. **D**  
In option D  $\frac{v^2}{c^2}$  is dimension less.

3. **B**  
4. **C**  
5. **C**  
6. **A**

Let that vector is  $\vec{C}$ . Then

$$\vec{C} = C\hat{C} = b\hat{a} \Rightarrow \vec{C} = \frac{b\hat{a}}{a} = \frac{5}{\sqrt{2}}(\hat{i} - \hat{j})$$

7. **A**



The angle which the resultant makes with  $P + Q$  will be  $(\alpha - \theta)$ .

$$\text{Thus } \tan(\alpha - \theta) = \frac{(P - Q) \sin 2\alpha}{(P + Q) + (P - Q) \cos 2\alpha}$$

$$\text{or } \frac{\sin(\alpha - \theta)}{\cos(\alpha - \theta)} = \frac{(P - Q) \sin 2\alpha}{(P + Q) + (P - Q) \cos 2\alpha}$$

$$\sin(\alpha + \theta) (P + Q) + (P - Q) \cos 2\alpha \sin(\alpha - \theta) = (P - Q) \sin 2\alpha \cos(\alpha - \theta)$$

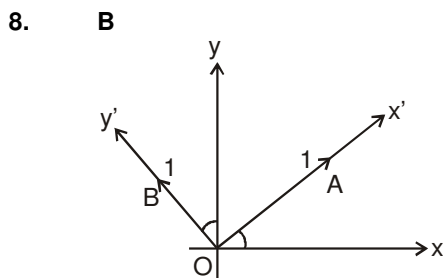
$$\sin(\alpha - \theta) (P + Q) = (P - Q) [\sin(2\alpha) \cos(\alpha - \theta) - \cos 2\alpha \sin(\alpha - \theta)]$$

$$\text{or } (P + Q) \sin(\alpha - \theta) = (P - Q) \sin(\alpha + \theta)$$

$$\text{or } P [\sin(\alpha + \theta) - \sin(\alpha - \theta)] = Q [\sin(\alpha + \theta) + \sin(\alpha - \theta)]$$

$$\text{or } P \times 2 \cos \alpha \sin \theta = Q \times 2 \sin \alpha \cos \theta$$

$$\therefore P \tan \theta = Q \tan \alpha$$



$$\hat{OA} = 1 \cos \theta \hat{i} + 1 \sin \theta \hat{j}$$

$$\hat{OB} = 1 \cos \theta \hat{j} + 1 \sin \theta (-\hat{i})$$

9. **A,B,C,D**

$$\text{Velocity} = \frac{\text{length}}{\text{time}}, \text{acc.} = \frac{\text{length}}{(\text{time})^2}$$

$$\Rightarrow \text{Length} = \frac{(\text{velocity})^2}{\text{acc.}}, \text{i.e., } L' = \frac{v^2}{a'} \text{ and } L = \frac{v^2}{a}$$

$$\Rightarrow \frac{L'}{L} = \left(\frac{v'}{v}\right)^2 \left(\frac{a}{a'}\right) = \left(\frac{\alpha^2}{\beta}\right) \frac{1}{\alpha\beta} = \frac{\alpha^3}{\beta^3}$$

$$\text{Now, } m' = \frac{F'}{a'} \text{ and } m = \frac{F}{a}$$

$$\Rightarrow \frac{m'}{m} = \frac{F' a}{F a'} = \frac{1}{\alpha\beta} \times \frac{1}{\alpha\beta} = \frac{1}{\alpha^2\beta^2}$$

$$\text{Time} = \frac{\text{vel.}}{\text{acc.}} \text{ i.e., } T' = \frac{v'}{a'} \text{ and } T = \frac{v}{a}$$

$$\Rightarrow \frac{T'}{T} = \frac{v' a}{v a'} = \frac{\alpha^2}{\beta} \frac{1}{\alpha\beta} = \frac{\alpha}{\beta^2}$$

$$\text{Momentum} = \text{mass} \times \text{velocity, i.e., } P' = m'v' \text{ and } P = mv$$

$$\Rightarrow \frac{P'}{P} = \frac{m' v'}{m v} = \frac{1}{\alpha^2\beta^2} \frac{\alpha}{\beta} = \frac{1}{\beta^3}$$

10. **A,B,C,D**

$$c^2 = a^2 + b^2 - 2ab \cos \theta \dots\dots\dots(1)$$

$$2a = b \dots\dots\dots(2)$$

$$\therefore c^2 = \frac{5a^2}{4} - a^2 \cos^2 \theta$$

Now check for A, B, C, D

11. **B,D**

$$\vec{A} = a_1\hat{i} + b_1\hat{j} \Rightarrow |A| = \sqrt{a_1^2 + b_1^2}$$

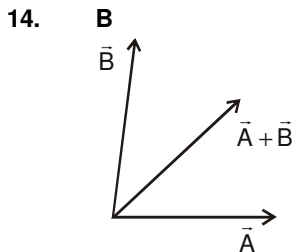
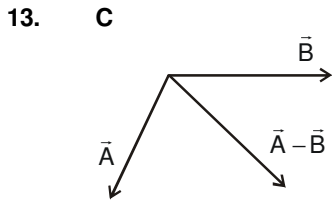
$$\vec{B} = a_2\hat{i} + b_2\hat{j} \Rightarrow |B| = \sqrt{a_2^2 + b_2^2}$$

$$A_x + B_x = a_1 + a_2 \text{ and } |A| + |B| = \sqrt{a_1^2 + b_1^2} + \sqrt{a_2^2 + b_2^2}$$

$$\Rightarrow \sqrt{a_1^2 + b_1^2} + \sqrt{a_2^2 + b_2^2} \geq a_1 + a_2$$

$$\Rightarrow |A| + |B| \geq A_x + B_x$$

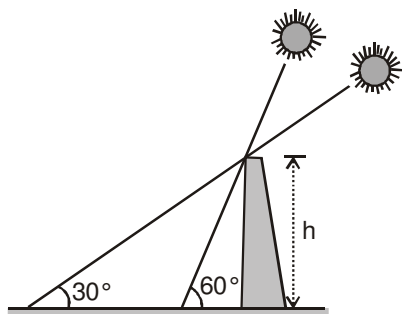
12. **A**  
'All rivers are sea, some streams are rivers, some streams are sea'.



15. **D**

**SECTION - C**

1. **150**



If  $h$  is the height of tower, then

$$h = \frac{d}{\cot \theta_2 - \cot \theta_1}$$

$$h = \frac{100\sqrt{3}}{\cot 30^\circ - \cot 60^\circ} = 50 \times 3 = 150$$

2. **10**

Let fundamental unit be  $v$ ,  $a$  and  $f$  :

$$v = [LT^{-1}], \quad \dots(1)$$

$$a = [LT^{-2}] \quad \dots(2)$$

and  $F = [MLT^{-2}] \quad \dots(3)$

Dividing (1) by (2),  $\frac{v}{a} = \frac{[LT^{-1}]}{[LT^{-2}]} = T$

$\therefore$  Unit of time ( $z$ ) =  $\frac{20\text{m/s}^2}{5\text{m/s}^2} = 4\text{sec}$

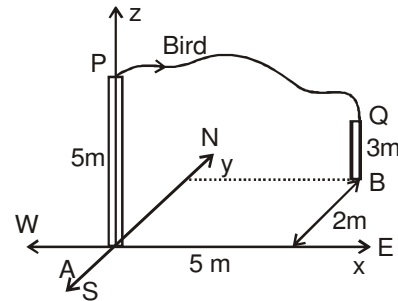
Dividing (3) by (2),  $\frac{F}{a} = M$

$\therefore$  Unit of mass ( $x$ ) =  $\frac{F}{a} = \frac{10}{5} = 2\text{kg}$

From (1)  $[L] = [vT] = 20 \times 4 = 80\text{m}$

$\Rightarrow \frac{y}{xz} = 10$

3. **6**



$$\vec{AP} = 5\hat{k}, \vec{AQ} = 5\hat{i} + 2\hat{j} + 3\hat{k}$$

Displacement of the bird =  $\vec{PQ} = \vec{AQ} - \vec{AP} = 5\hat{i} + 2\hat{j} - 2\hat{k}$

$$PQ = \sqrt{5^2 + 2^2 + 2^2} = \sqrt{33}\text{m} \approx 6$$

4. **6**

Subjective

$$\vec{AB} + \vec{BC} + \vec{CD} = \vec{AD} \quad \dots(1)$$

$$\vec{AF} + \vec{FE} + \vec{ED} = \vec{AD} \quad \dots(2)$$

From diagram

$$\vec{AC} = \vec{AB} + \vec{CD}$$

$$\vec{AC} = \vec{ED} + \vec{FE} \quad \dots(3)$$

$$\vec{AE} = \vec{AF} + \vec{FE} = \vec{CD} + \vec{BC} \quad \dots(4)$$

keeping the value (3) & (4) in (1) & (2) & adding (1) & (2)

$$\vec{AB} + \vec{AE} + \vec{AF} + \vec{AC} = 2\vec{AD}$$

$$\vec{AD} = 2\vec{AO}$$

$$\vec{AB} + \vec{AC} + \vec{AD} + \vec{AF} + \vec{AE} = 3\vec{AD} = 6\vec{AO}$$

5. **7**

The length of the vector is not changed by the rotation of the coordinate axes.

$$\sqrt{(n+1)^2 + 1^2} = \sqrt{n^2 + 3^2} \quad \text{or } n^2 + 2n + 2 = n^2 + 9$$

$$\text{or } 2n = 7$$

## CHEMISTRY

### SECTION - A

1. **C**
2. **A**
3. **A**
4. **B**
5. **C**  
Atomic Mass is a unitless quantity as implied from its definition.
6. **B**  
It is in decreasing order of radii.
7. **C**  
 $PV = nRT$   
 $V = \frac{nRT}{p}$   
 $V = \frac{2 \times 0.0821 \times 273}{1} = 44.8 \text{ L}$
8. **C**  
Molecular Mass of  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O} = 249.5$   
 $n = \frac{499}{249.5} = 2 \text{ moles}$   
1 mole  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  contains O atoms = 9 moles  
2 mole  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  contains O atoms =  $9 \times 2 = 18 \text{ moles}$
9. **ABCD**
10. **ABC**  
Molecular Mass of  $\text{NH}_3 = 17$   
 $n = \frac{1.7}{17} = 0.1 \text{ mole}$   
no. of atoms =  $0.1 \times 4 \times N_A = 0.4 \times N_A$   
At STP 1 mole gas occupies = 22.7 L  
So, 0.1 gas will occupy = 2.27 L  
1 mole  $\text{NH}_3$  contains electron =  $7 + 3 \times 1) N_A = 10 \times N_A$   
So, 0.1 mole  $\text{NH}_3$  will contain =  $(10 \times N_A) \times 0.1 = N_A \text{ Ans.}$
11. **ABC**  
22.7 L at STP = 1 mole  
5.675 L at STP =  $\frac{1}{22.7} \times 5.675 = 0.25 \text{ mole}$   
1 gm He =  $\frac{1}{4} \text{ mole} = 0.25 \text{ mole}$   
0.5 gm H atom =  $\frac{0.5}{1} \text{ mole H atom}$   
 $= 0.5 \times \frac{1}{2} \text{ mole H}_2 \text{ molecule} = 0.25 \text{ mole}$   
11 gm  $\text{CO}_2 = \frac{11}{44} \text{ mole CO}_2 = 0.25 \text{ mole}$   
2 gm  $\text{H}_2 \text{ gas} = \frac{2}{2} = 1 \text{ mole}$

12. **B**
13. **B**
14. **B**
15. **A**

### SECTION C

1. **0050 elements**  
 $n = 9$   
 $9s^2, 9p^6, 8d^{10}, 7f^{14}, 6g^{18}$   
 $2 + 6 + 10 + 14 + 18 = 50 \text{ elements}$   
(5  $n^{22}$  is not possible)
2. **0016**  
 $n = 4$   
 $\ell = 0 \quad s \rightarrow 1 \text{ orbitals}$   
 $1 \quad p \rightarrow 3 \text{ orbitals}$   
 $2 \quad d \rightarrow 5 \text{ orbitals}$   
 $3 \quad f \rightarrow 7 \text{ orbitals}$   
Total orbitals =  $1 + 3 + 5 + 7 = 16$
3. **2.2 (2200)**  
 $Zs = 55$   
 $\sigma = (0.85 \times 8) + 46 \times 1.0 = 52.80$   
 $z^* = z - \sigma$   
 $= 55 - 52.8 = 2.2$   
Answer to be marked as  $2.2 \times 1000 = 2200 \text{ Ans.}$
4. **0016**  
22.7 L at STP = 1 mole  
 $45.4 \text{ L} = \frac{1}{22.7} \times 45.4 = 2 \text{ moles}$   
super heated steam is nothing but  $\text{H}_2\text{O}$ .  
So, no. of neutrons in 1 mole  $\text{H}_2\text{O}$   
 $= [(2 \times 0 + (8 \times 1))] \text{ moles}$   
So, 2 mole  $\text{H}_2\text{O}$  contains =  $8 \times 2 = 16 \text{ mole neutrons}$
5. **4000**  
1 tonne = 1000 kg  
Total molar of Urea manufactured every day  
 $= \frac{12 \times 1000 \times 1000 \text{ gm}}{60 \text{ gm/mol}} = 2 \times 10^5 \text{ moles}$   
mol. wt. of Urea  $\text{NH}_2\text{CONH}_2 = 60$   
10,000 moles of Urea = Rs 20,000 Profit  
So, for 1 mole Urea = Rs 2 profit  
for  $2 \times 10^5$  moles Urea = Rs.  $2 \times 2 \times 10^5$  profit  
 $= \frac{4 \times 10^5}{100} = 4,000$